

**THE
PERVASIVE
PLASMA**

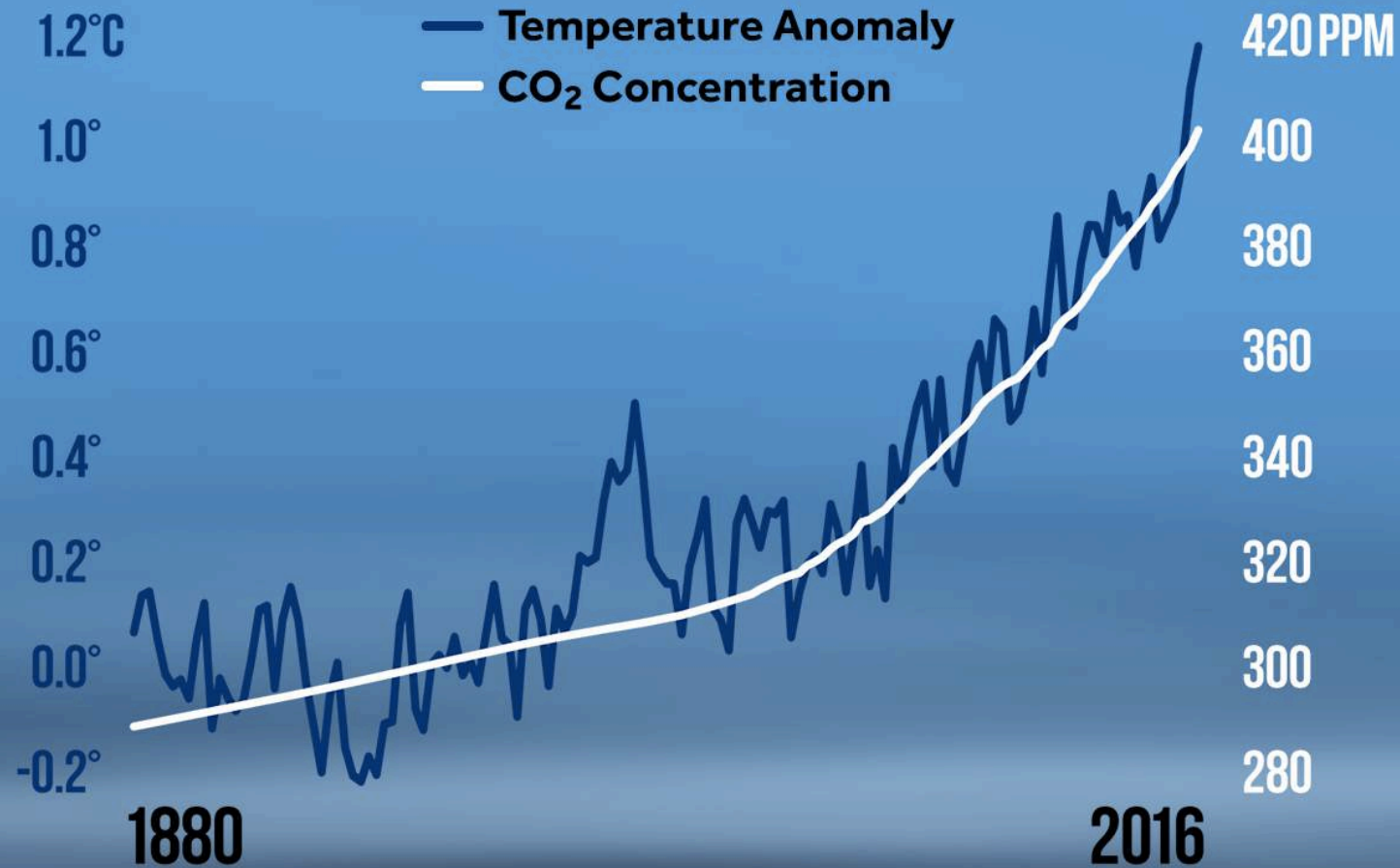
Plasma Processes for Carbon-free Energy

P. I. John

Vikram Sarabhai Oration
Shri Vaishnav Vidyapeeth
Vishwavidyalaya, Indore
18 December 2018

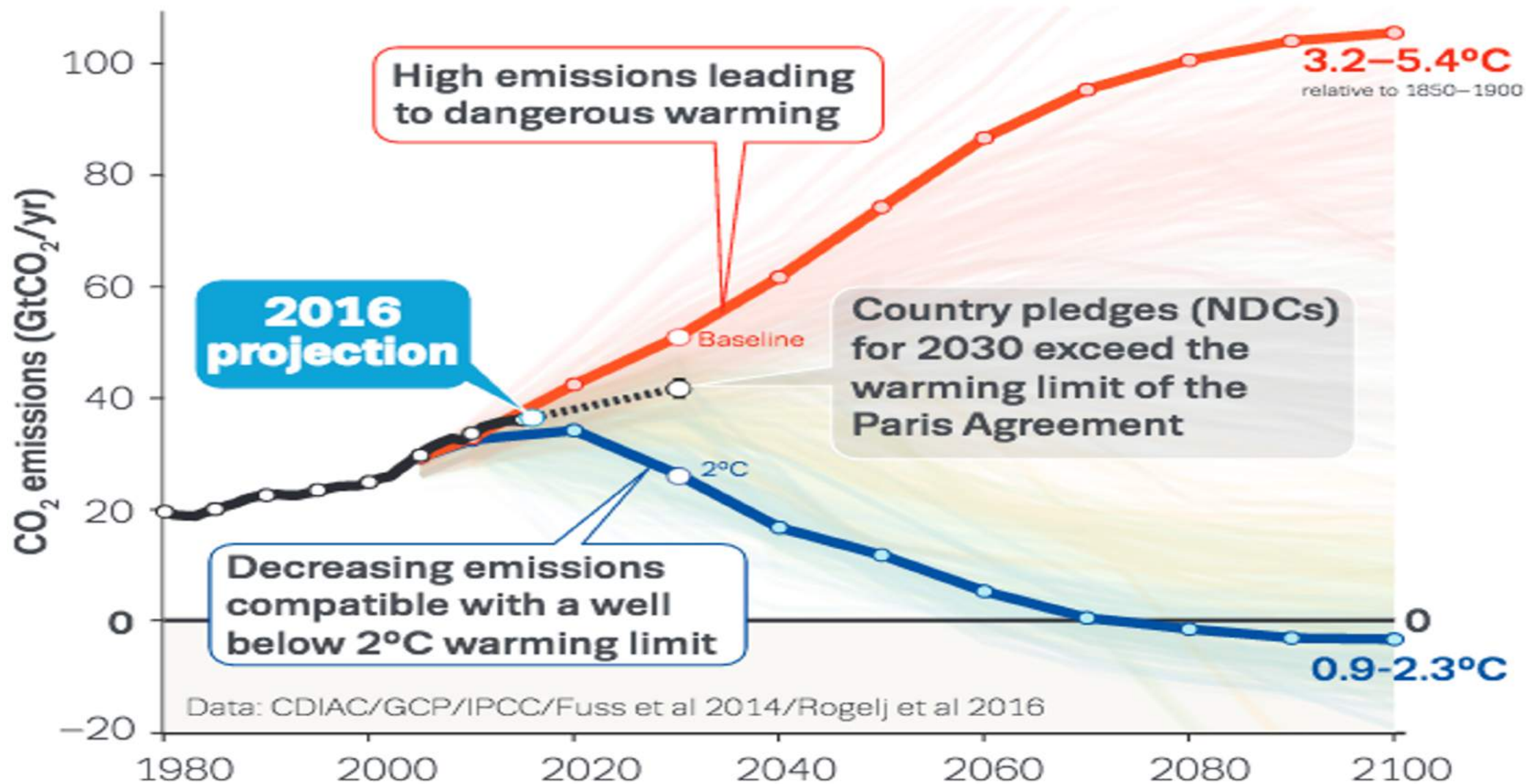
I

Global Temperature and Carbon Dioxide



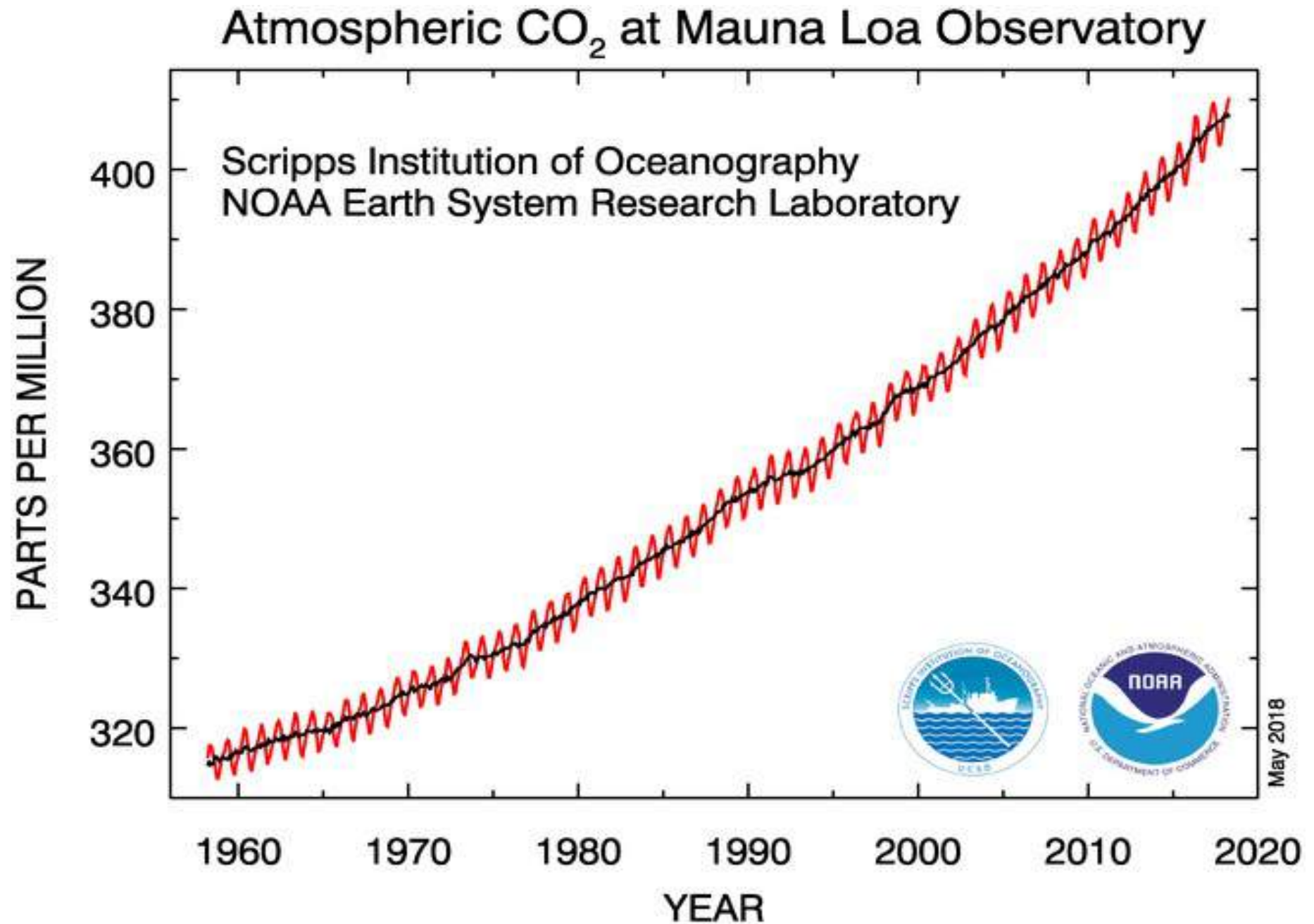
Global temperature data averaged and adjusted to early industrial baseline (1881-1910).
Source: NASA GISS, NOAA NCEI, ESRL

CLIMATE  CENTRAL



Source: Earth Syst. Sci. Data, 8, 1–45, 2016 Global Carbon Budget 2016 Corinne Le Quéré et. al.

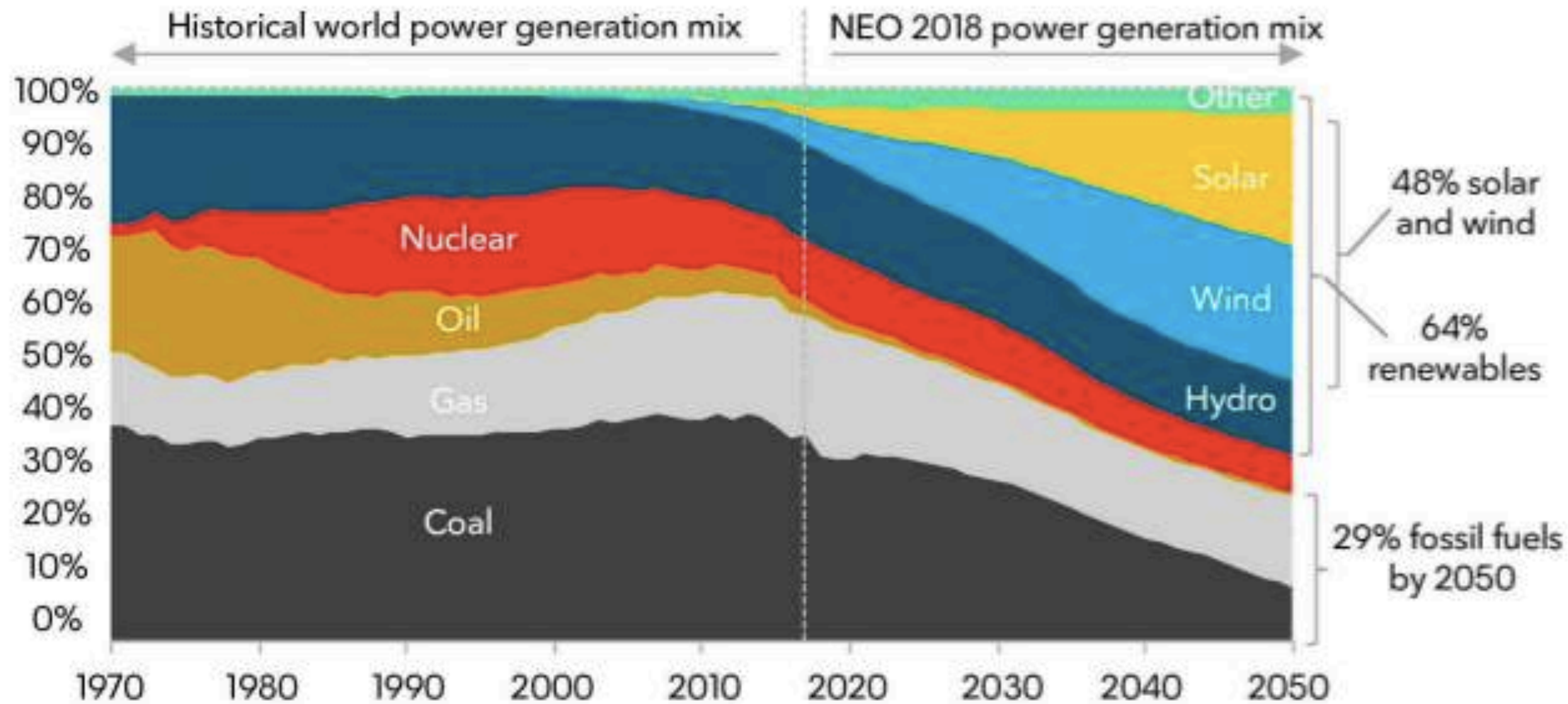
Above
410 ppm
for the first
time in
April 2017



Global Warming of 1.5°C

An IPCC special report on the impacts of global warming of 1.5°C above pre-industrial levels and related global greenhouse gas emission pathways, in the context of strengthening the global response to the threat of climate change, sustainable development, and efforts to eradicate poverty.

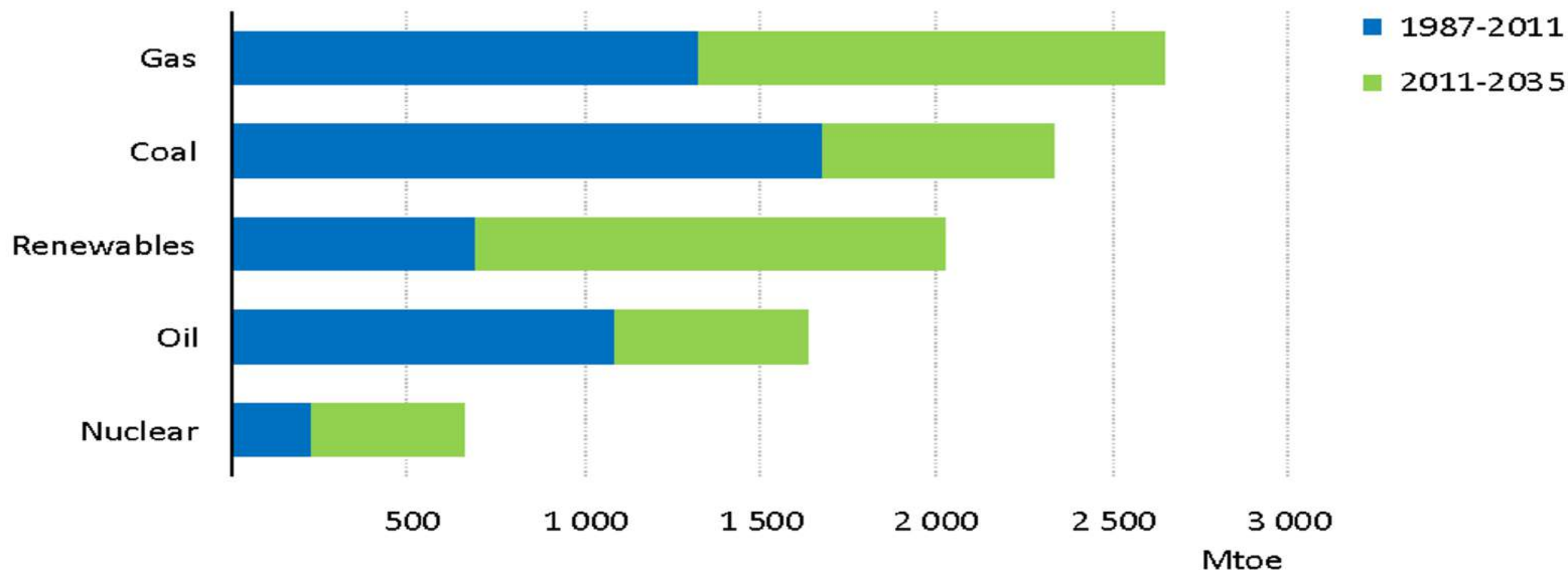
Power generation mix



A mix that is slow to change

WORLD
ENERGY
OUTLOOK
2013

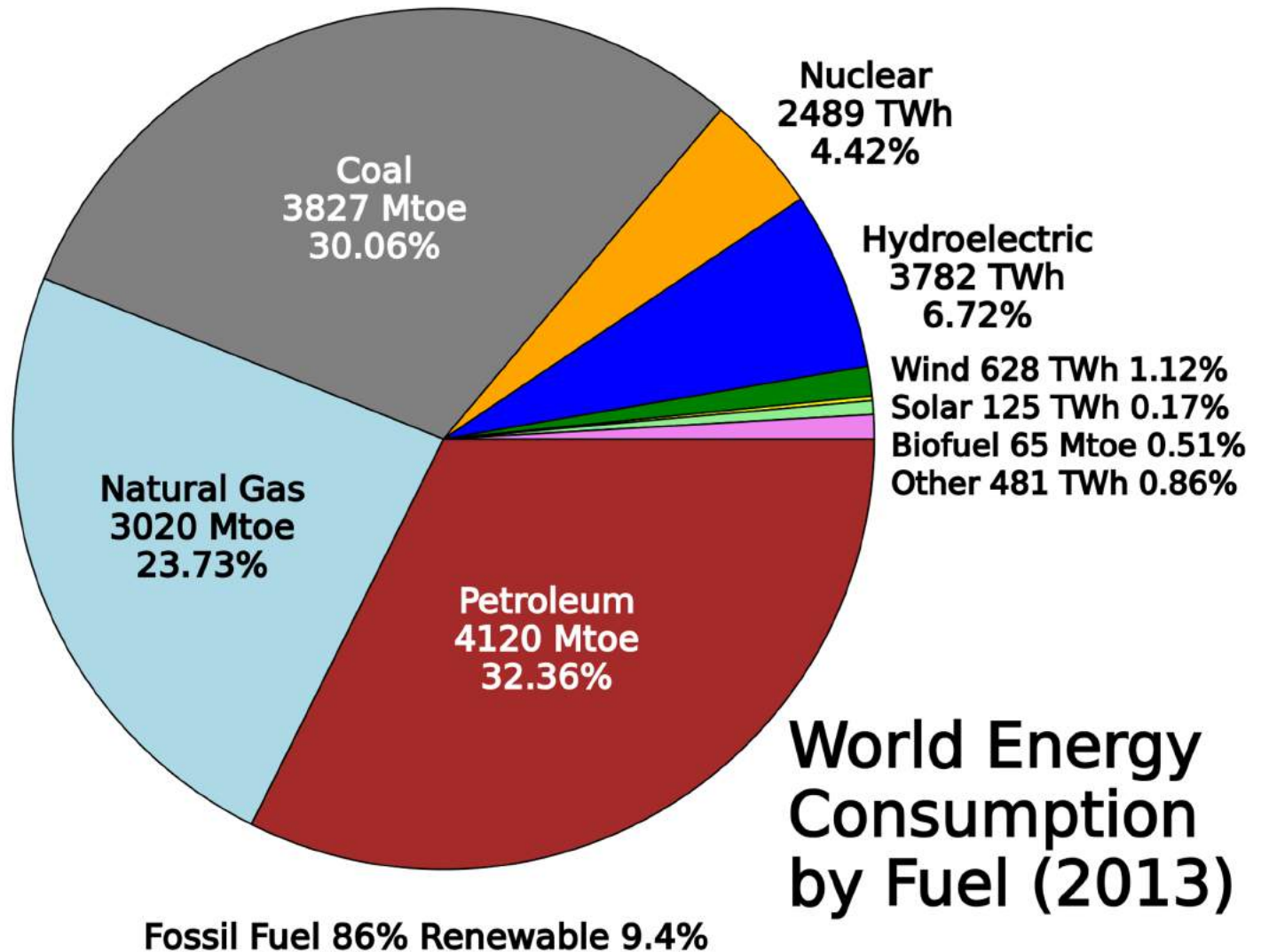
Growth in total primary energy demand



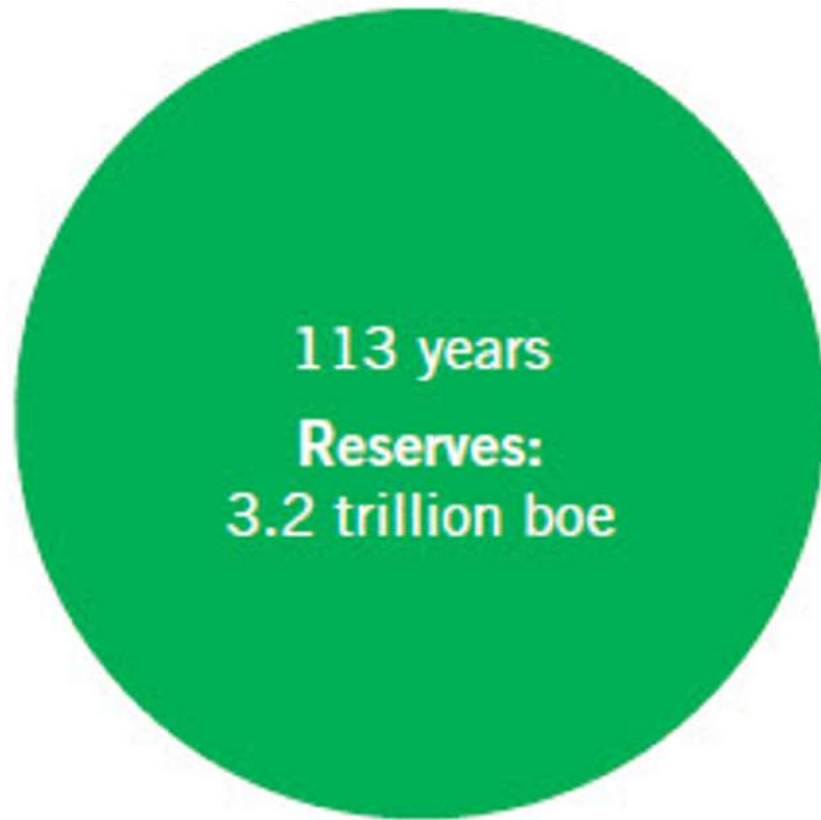
Today's share of fossil fuels in the global mix, at 82%, is the same as it was 25 years ago; the strong rise of renewables only reduces this to around 75% in 2035

85% of the world energy demand is met by non-renewable fossil fuels

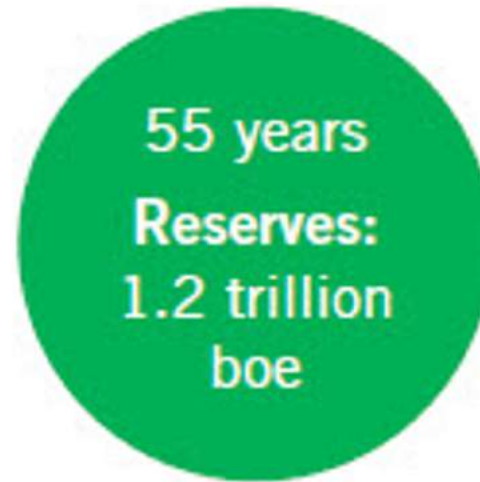
Accessibility
Broad applications
Transportation fuel
Portability
Storage



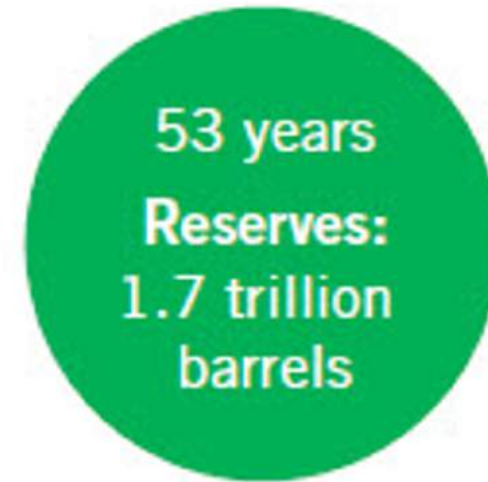
Bubbles show proved reserves in barrels of oil equivalent (boe) with bubble size expressed as the number of years of production remaining based on estimated production in 2013.



Coal



Gas



Oil

Source: BP, 2014. *Statistical Review of World Energy*.

OPTION 1: DECARBONIZE FUEL

**DECARBONIZATION: PROCESS OF
REDUCING CARBON INTENSITY IN ENERGY
UTILIZATION**

EVOLUTION OF THE HYDROGEN-TO-CARBON RATIO IN THE WORLD'S PRIMARY FUEL MIX

Click To Enlarge
Click Again To
Zoom In And Out



OIL CO₂ = 161

NATURAL GAS CO₂ = 117

METHANE: H/C = 4

OIL: H/C = 2

COAL CO₂ = 206

COAL: H/C = 0.5

WOOD CO₂ = 324

WOOD: H/C = 0.1

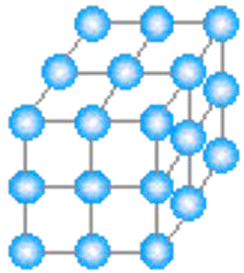
PATH OF
H/C RATIO EVOLUTION

KEY:
● - HYDROGEN (H)
● - CARBON (C)

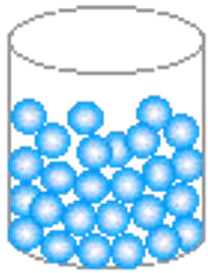
CO₂ = Pounds per million BTU (One BTU heats one pound of water one degree Farenheit.)

**CAN PLASMA BASED
TECHNOLOGIES HELP IN
DECARBONIZING FUELS?**

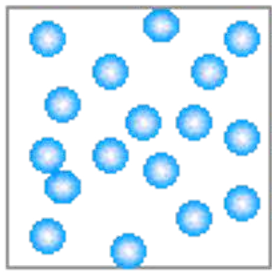
States of Matter



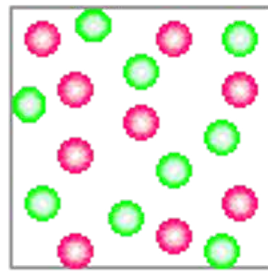
SOLID



LIQUID



GAS



PLASMA



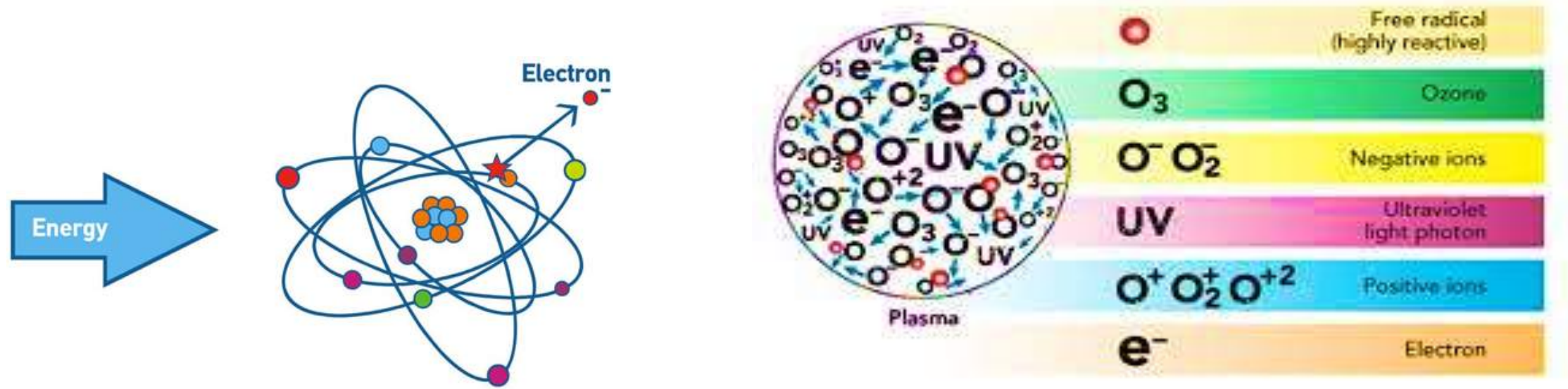
Cloud of electrically charged particles.

Coulomb interaction between the particles control dynamics.

Leads to collective effects like oscillations, waves, instabilities and self organization.

Most of the matter in the Universe exists as Plasma.

Sun, solar wind, stars, galactic space are all made of plasma.



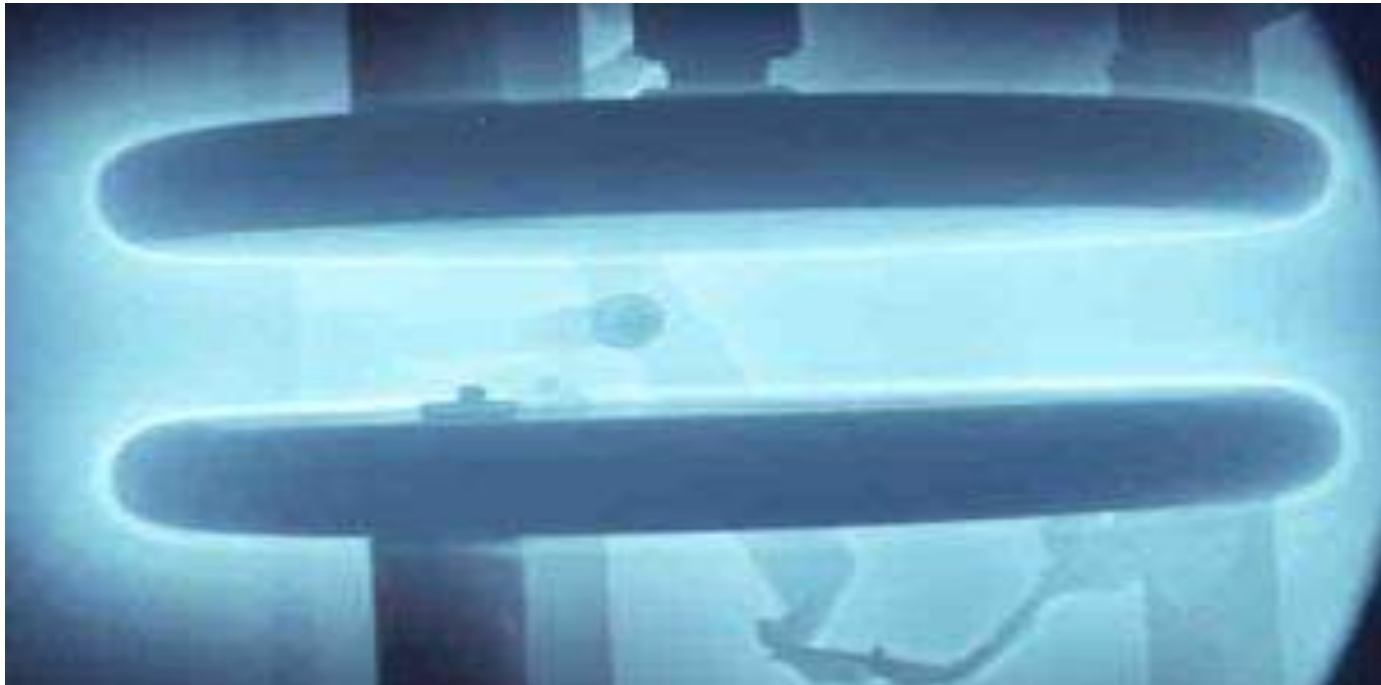
Energetic electrons knock off electrons of an atom and ionize it.

If the gas is situated in an electric field, the product electrons can be accelerated to gain energy and produce more ionization.

Repeated many times, this process leads to conversion of the gas into a plasma.

Energy flows from the electric field to the electrons and through collisions, to ions and neutrals.

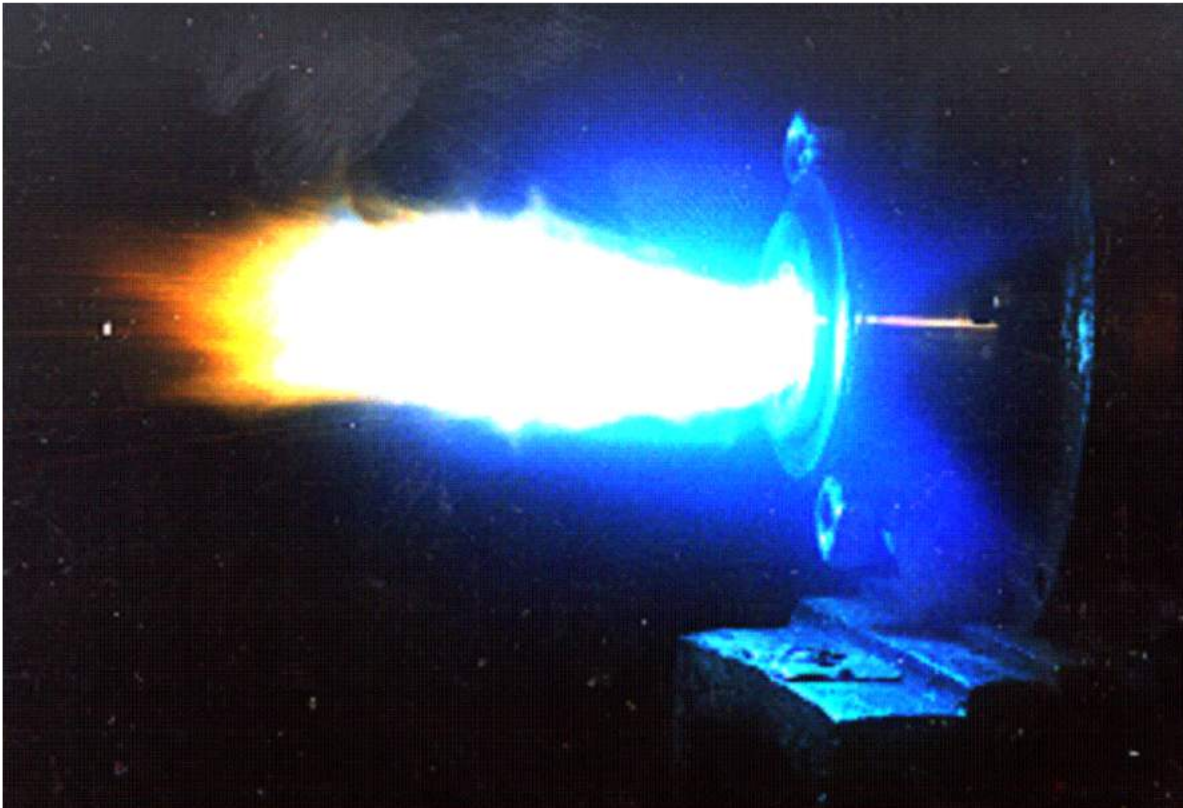
At low pressure, with infrequent collisions, the electrons are hot whereas the ions and neutrals are cold. The plasma is called non-equilibrium or cold plasma.



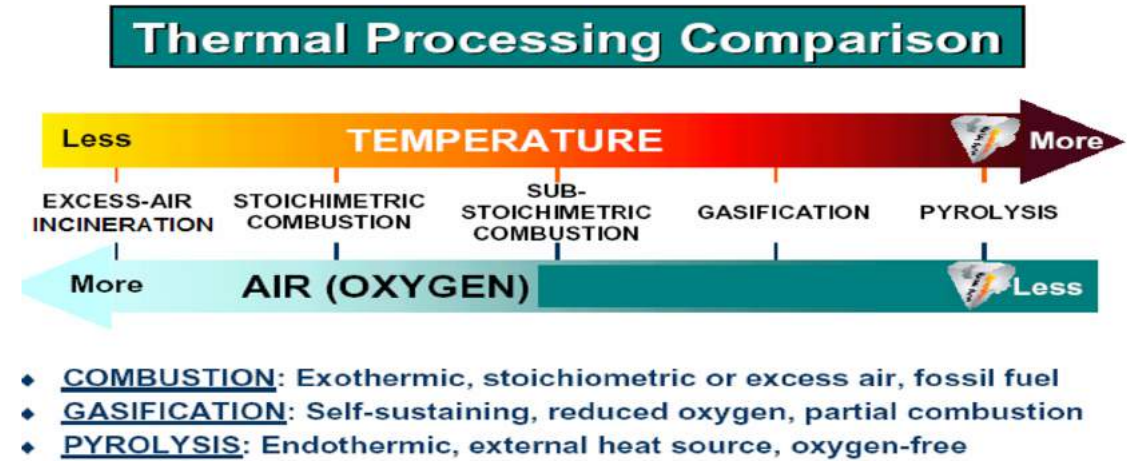
Plasma environment has high chemical reactivity because of the presence of free electrons.

Exotic Chemistry!

At high pressure, electrons and ions thermalize, producing hot plasma with temperatures of the order of 20,000 Deg K



PLASMA TORCH



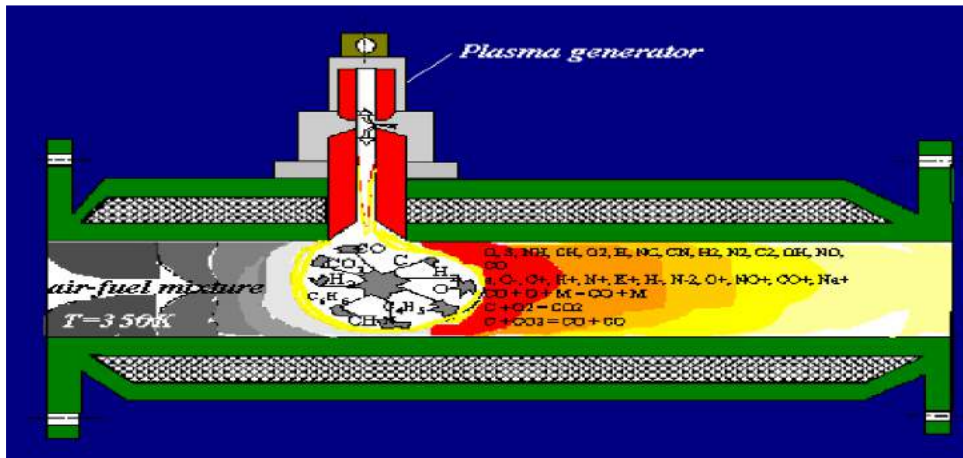
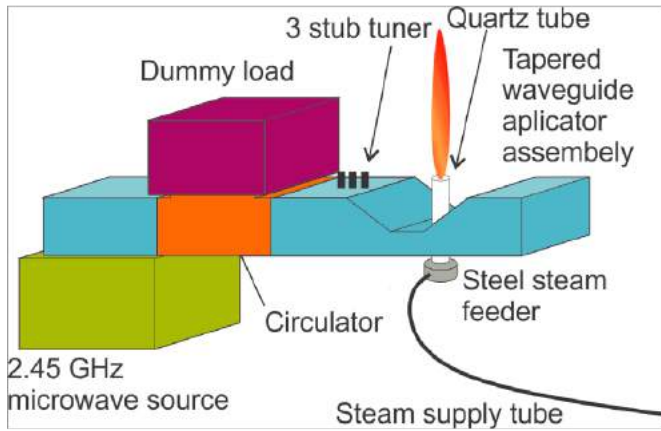
Heat causes organic material to disintegrate forming fragments promoted by the high chemical reactivity of the plasma environment. Most likely products are H, CH₄, C₂H₆, CO etc. 10% remain as Char/Ash

PLASMA TECHNOLOGIES FOR FUEL DECARBONIZATION

GASIFICATION

50-100 micron particles fragment into 5-10 micron particles and are volatilized to produce CO, CO₂, H₂, N₂, CH₄, C₆H₆ etc.

5kW Plasma Coal Gasification activity at FCIPT



PLASMA TECHNOLOGIES FOR FUEL DECARBONIZATION

FUEL REFORMING

Partial Oxidation: $\text{CH}_4 + 0.5\text{O}_2 = \text{CO} + 2\text{H}_2$ (-35.6 kJ/mol)

Water Gas Shift: $\text{CO} + \text{H}_2\text{O} = \text{CO}_2 + \text{H}_2$ (-41.2)

Methane Oxidation: $\text{CH}_4 + 2\text{O}_2 = \text{CO}_2 + 2\text{H}_2\text{O}$ (-802.2)

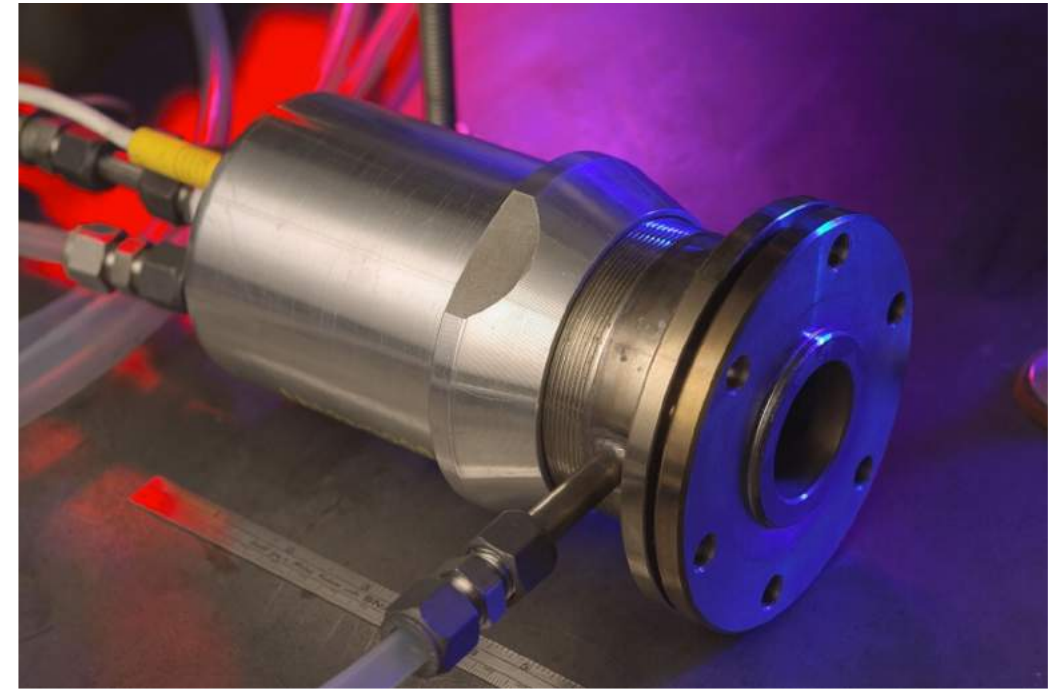
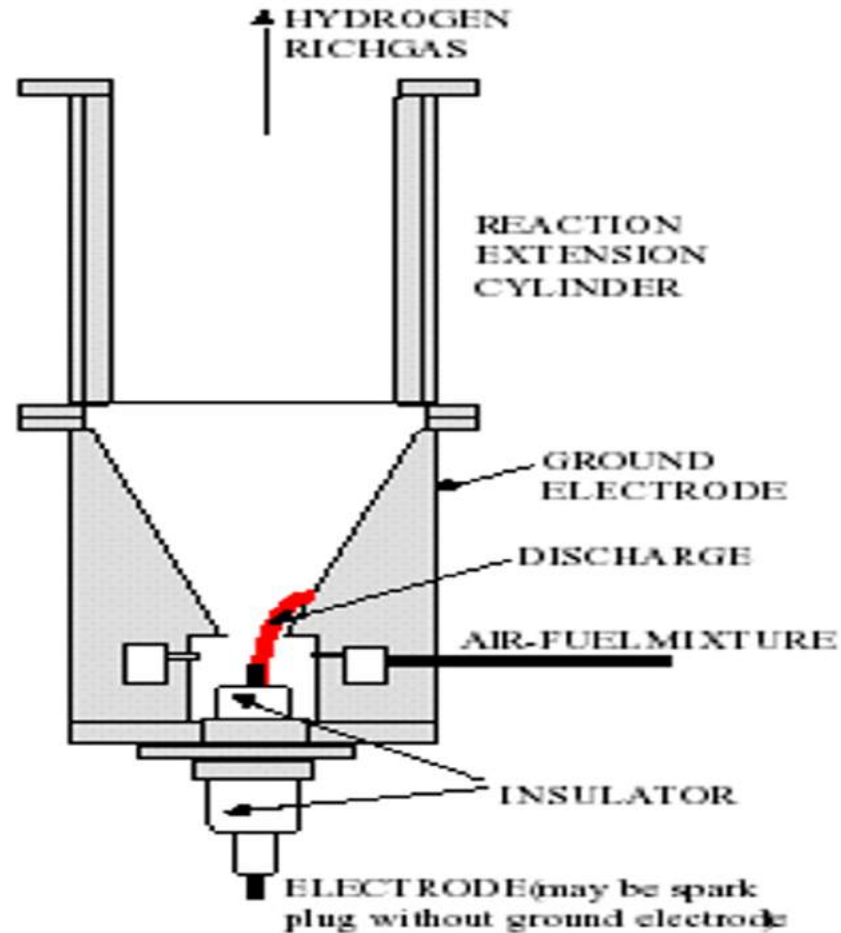
CO Oxidation: $\text{CO} + 0.5\text{O}_2 = \text{CO}_2$ (-283)

CO Methanation: $\text{CO} + 3\text{H}_2 = \text{CH}_4 + \text{H}_2\text{O}$ (-206.2)

CO₂ Methanation: $\text{CO}_2 + 4\text{H}_2 = \text{CH}_4 + 2\text{H}_2\text{O}$ (-165)

- **Steam Reforming:**
Convert Methane into H₂
- At high temperature and in the presence of metal based catalyst (Nickel), steam reacts with Methane to yield CO and H₂ through the following reactions:
 - $\text{CH}_4 + \text{H}_2\text{O} = \text{CO} + 3\text{H}_2$
 - $\text{CO} + \text{H}_2\text{O} = \text{CO}_2 + \text{H}_2$
- **CO₂ must be sequestered**

Compact, lightweight, fast, controllable, arc plasma based reformer for onboard fuel reforming can be a viable transportation technology



ArvinMeritor

70% conversion efficiency,

Power 50-300 W

H₂ flow rate 30-50 liters/min

Volume 2 liters

Direct Decarbonization

Laurent Fulcheri, Mines - Paris Tech, France

Methane reforming produces 4 tonnes of CO₂ per tonne of H₂

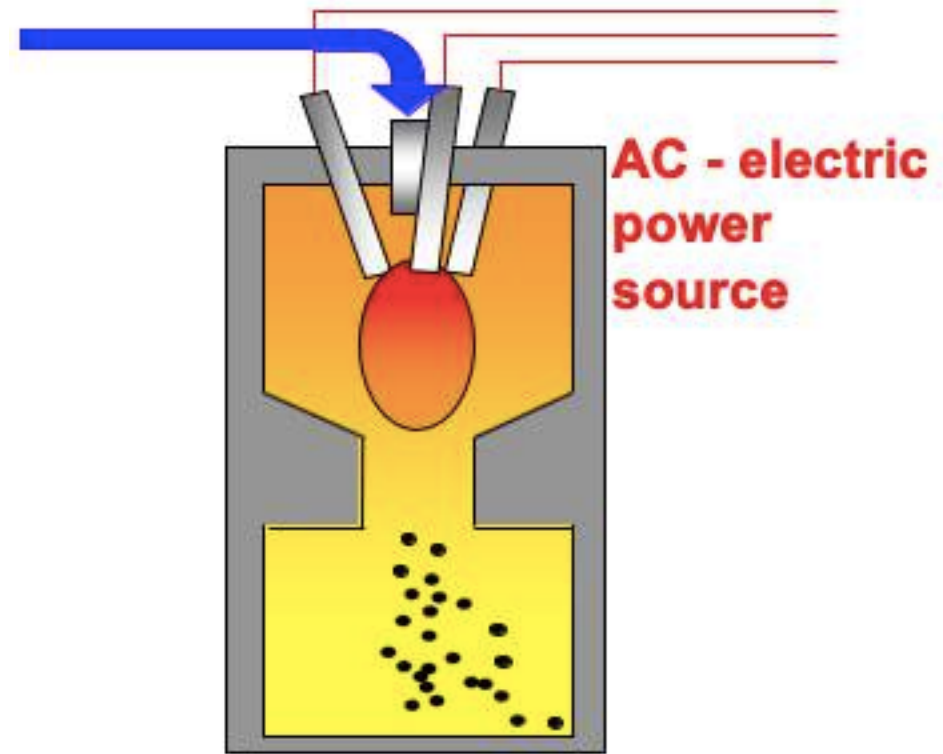
Alternative, dry process: $\text{CH}_4 \rightarrow \text{C} + 2\text{H}_2$ (75:6kJ/mol)

High temperature (>1200 °C).

Byproduct is valuable carbon black.

Zero direct CO₂

\$2/kg Hydrogen DOE objective

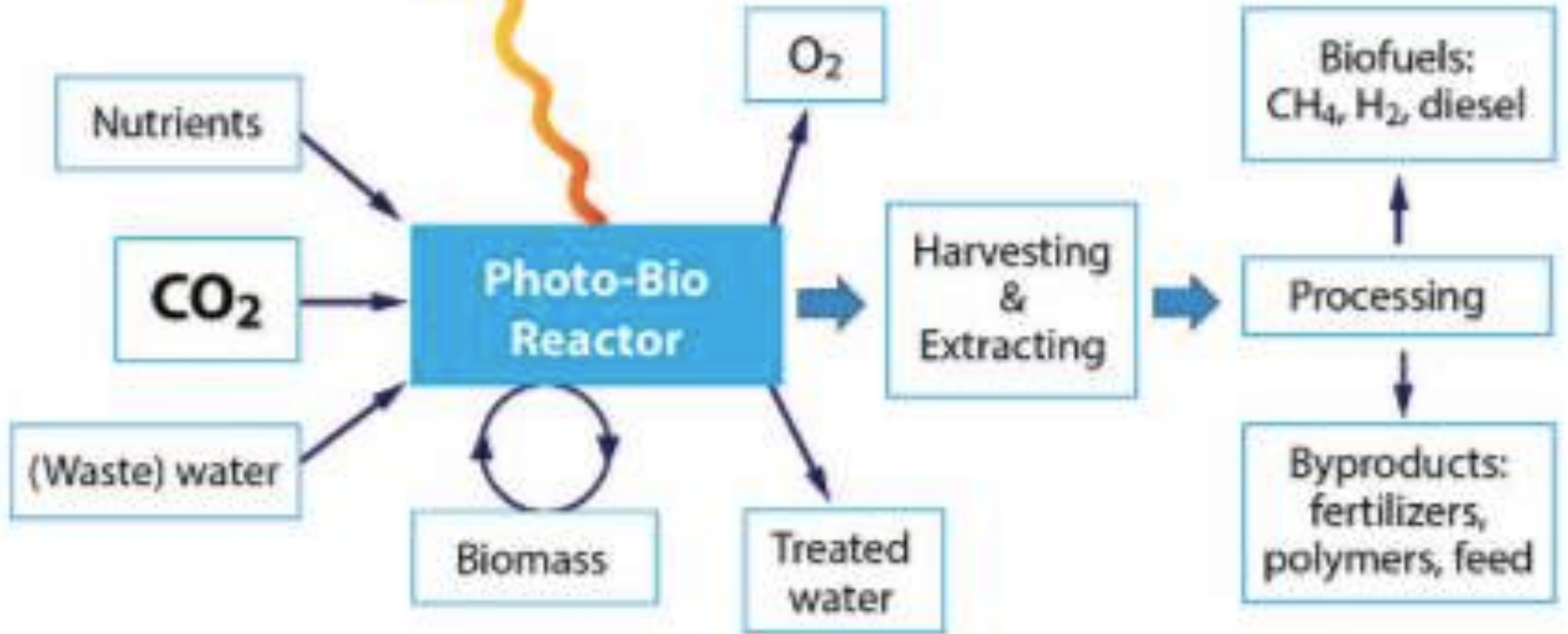


OPTION 2

DECARBONIZING THE ATMOSPHERE

**CAN WE REMOVE CO₂ FROM THE
ATMOSPHERE AND CONVERT IT INTO
FUEL?**

PHOTOSYNTHESIS

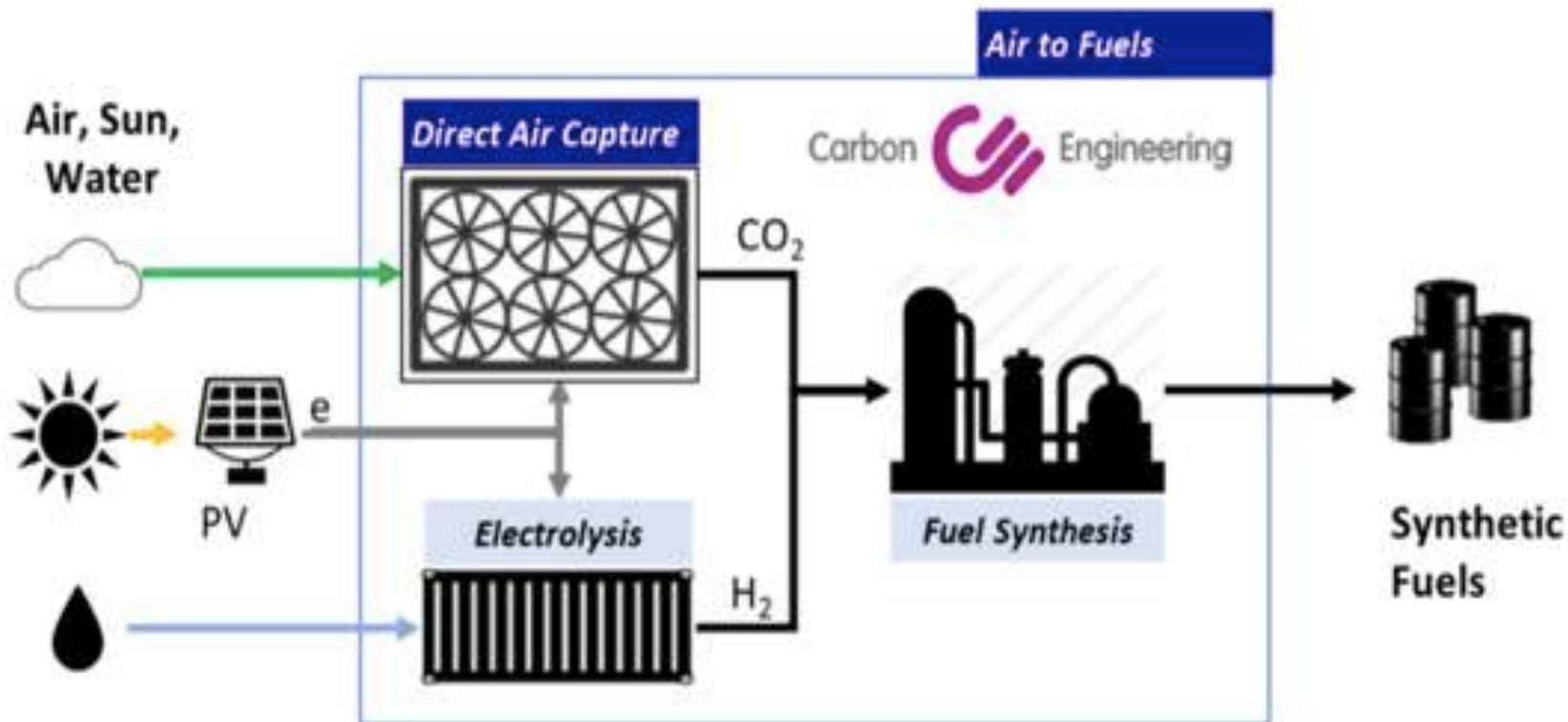




AIR TO FUEL (A2F) PROCESS

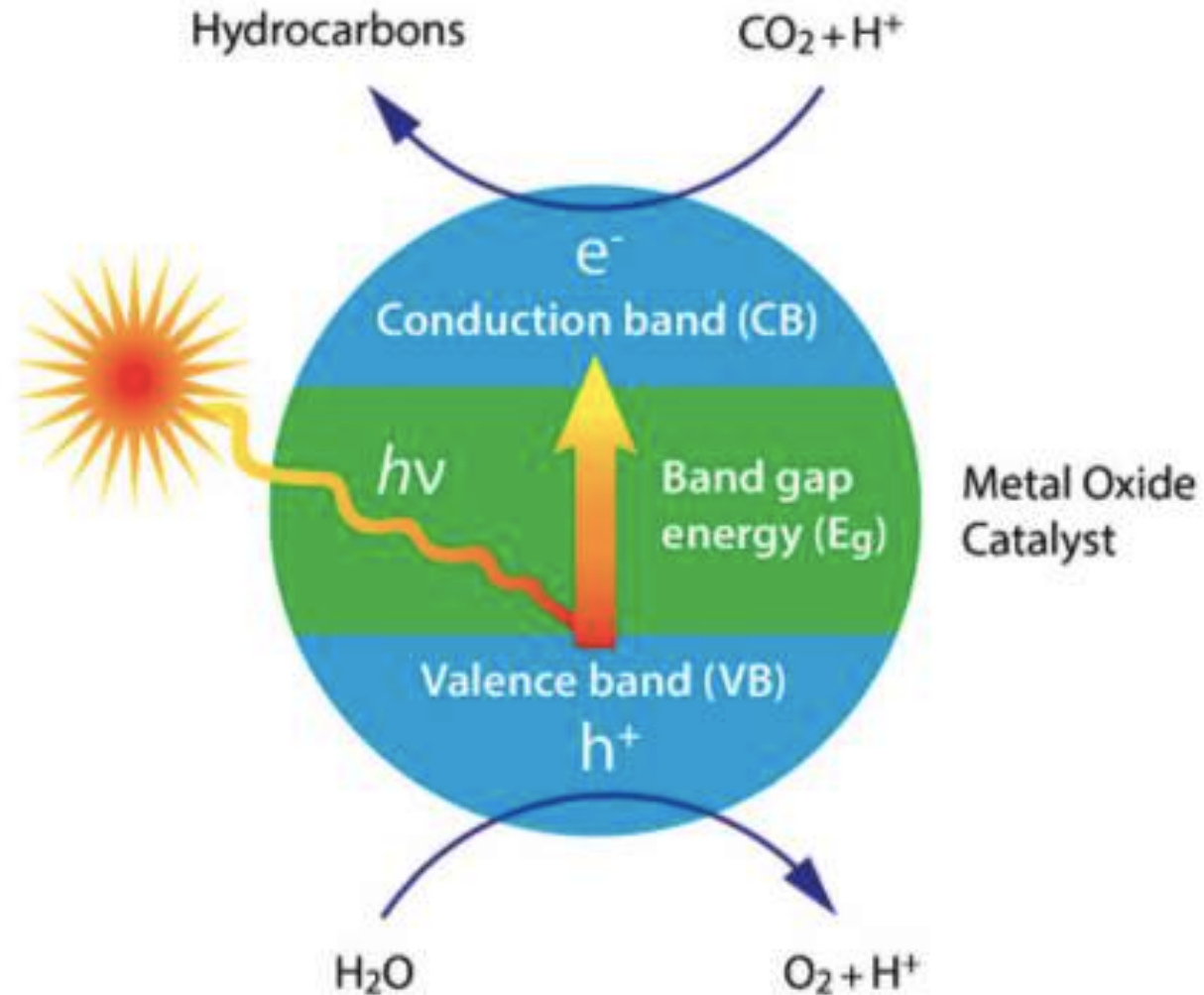
Cost: US\$ 94 and \$ 232 per tonne.

CE has a pilot in Canada since 2015

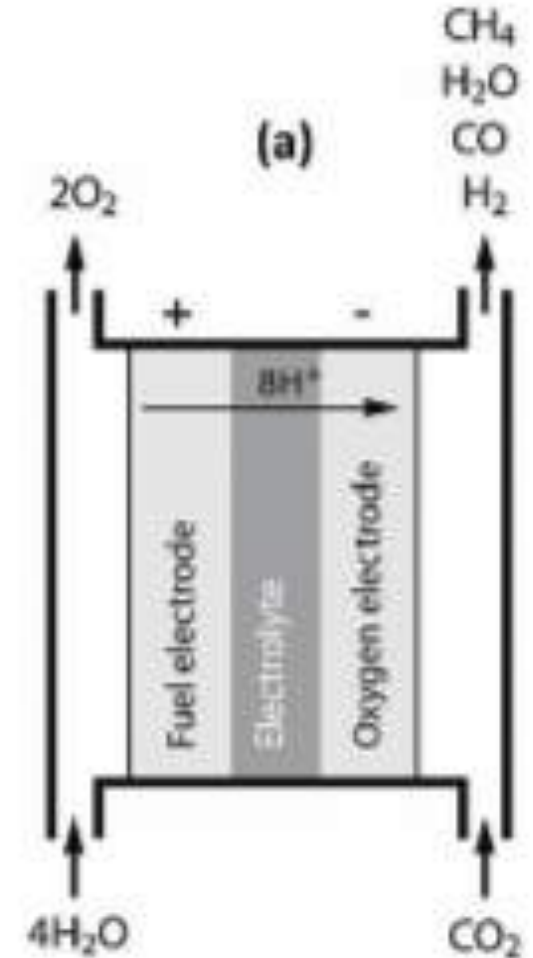


Source: The Economist
Jun 7th 2018

PHOTOCATALYTIC REDUCTION



ELECTROCHEMICAL REDUCTION



	Use of rare earth metals	Renewable energy	Turnkey process	Conversion and yield	Separation step needed	Oxygenated products (e.g. alcohols, acids)	Investment cost	Operating cost	Overall flexibility
Traditional catalysis	Yes		No	High	Yes	Yes	Low	High	Low
Catalysis by MW-heating		Indirect						Low	Low
Electro-chemical	Yes	Indirect	No ^b	High	Yes ^c	Yes	Low	Low	Medium
Solar thermo-chemical	Yes	Direct	NA	High	No	No	High	Low	Low
Photo-chemical	Yes	Direct ^a	Yes	Low	Yes	Yes	Low	Low	Low
Biochemical	No	Direct ^a	No	Medium	Yes ^d	Yes	High /low	High	Low

Source: R. Snoeckx and A.Bogaerts, Chem. Soc. Rev., 2017,46, 5805

PLASMA DISSOCIATION OF CO₂

WHY PLASMA?

Dutch Institute for Fundamental
Energy Research The Netherlands

Institut für Plasmaforschung
Universität Stuttgart, Germany

Department of Physics and Astronomy,
West Virginia University, USA

Research group PLASMANT, University
of Antwerp, Belgium

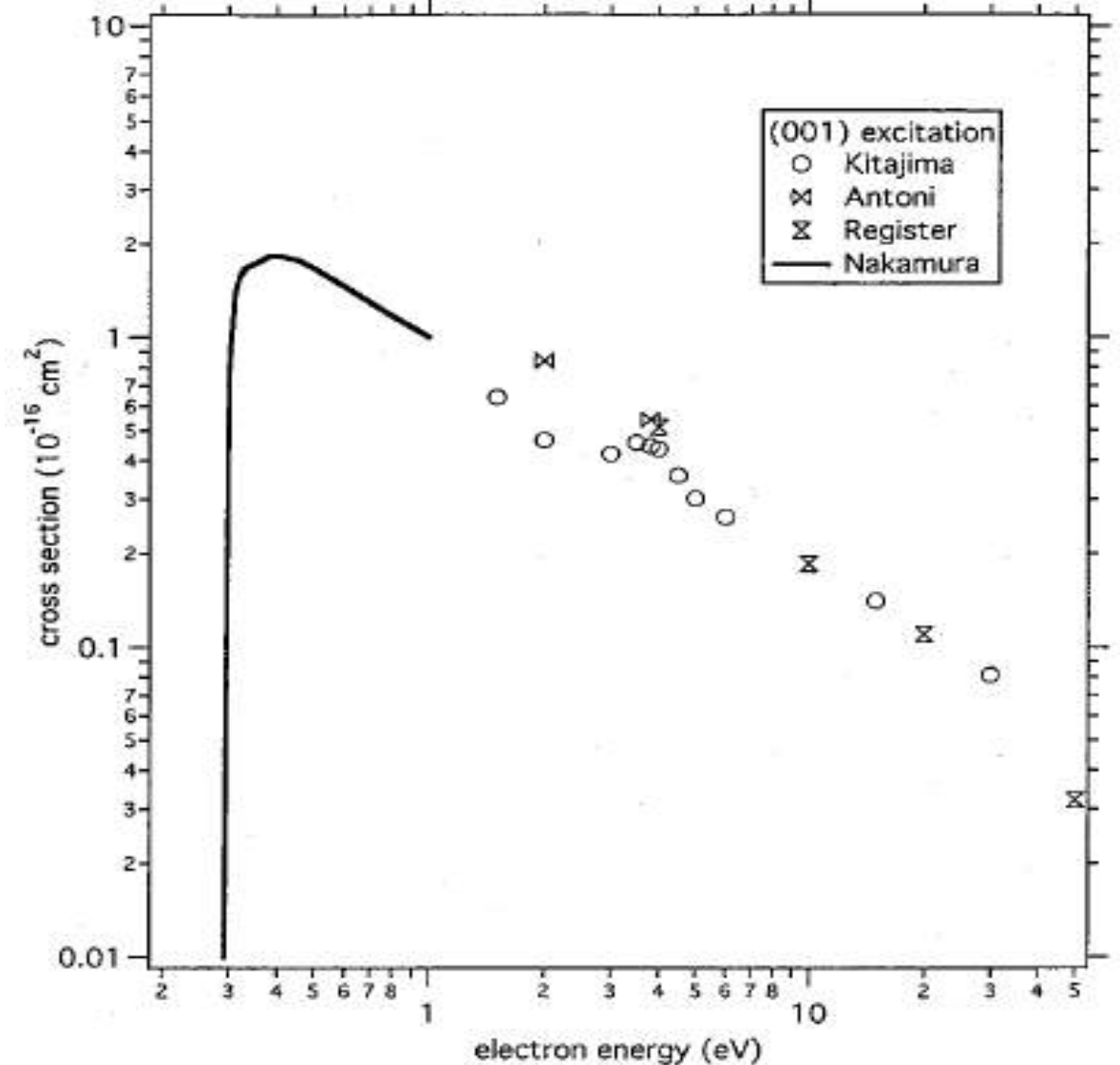
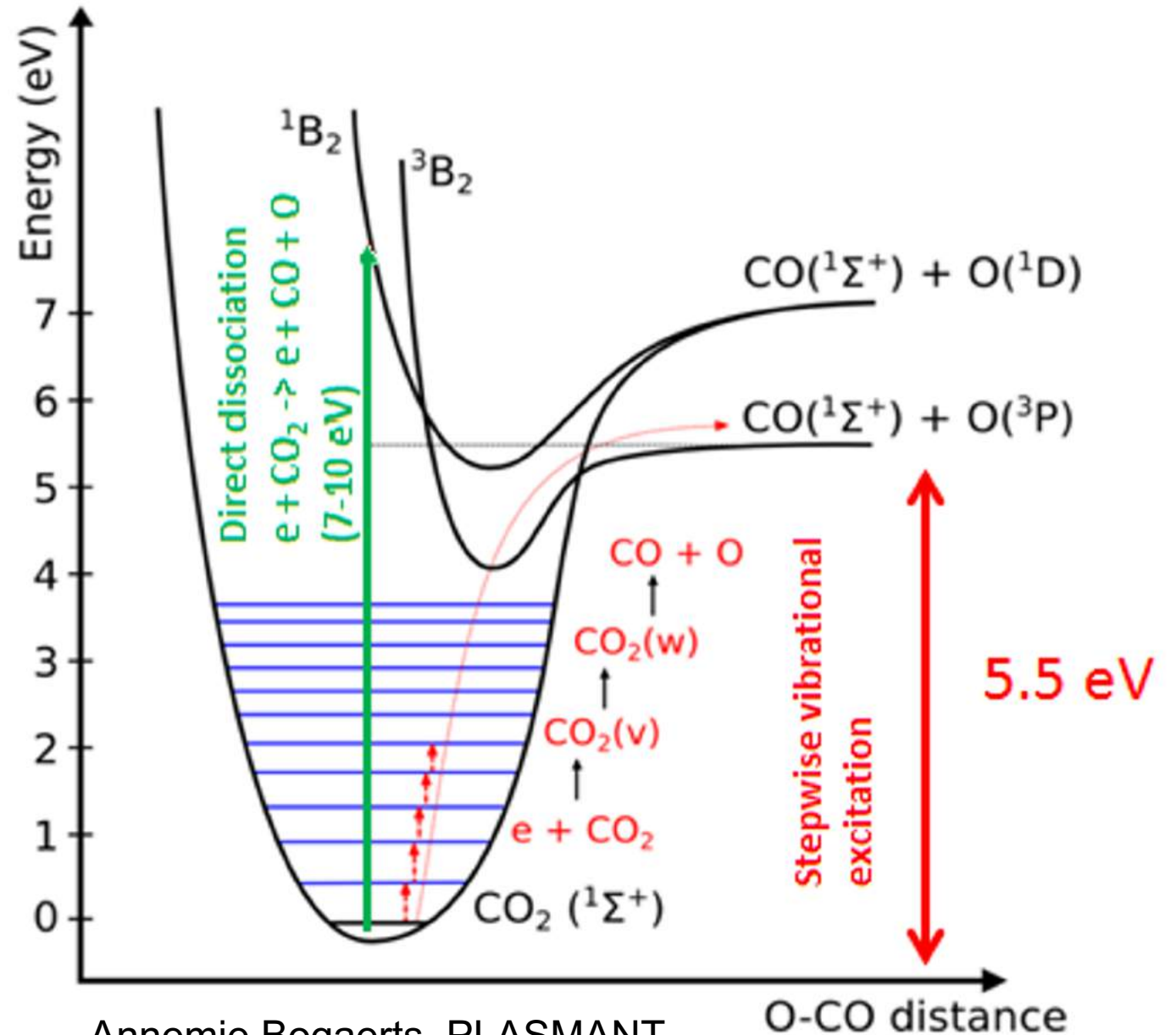
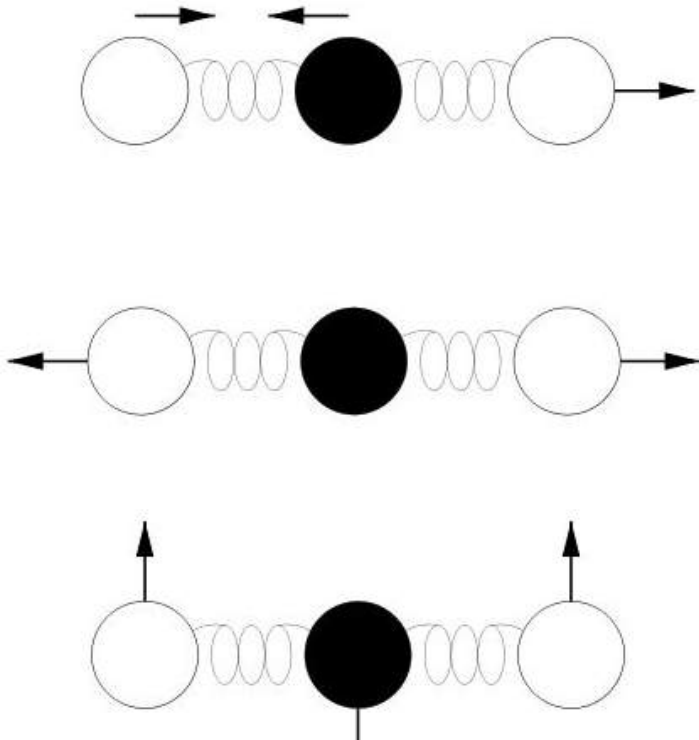


FIG. 7. Cross sections for the electron-impact excitation of the vibrational state (001) of CO₂. Comparison of the beam experiments by Kitajima *et al.*,³² Antoni *et al.*,³³ and Register *et al.*,²⁶ and the swarm result of Nakamura³¹ is shown.

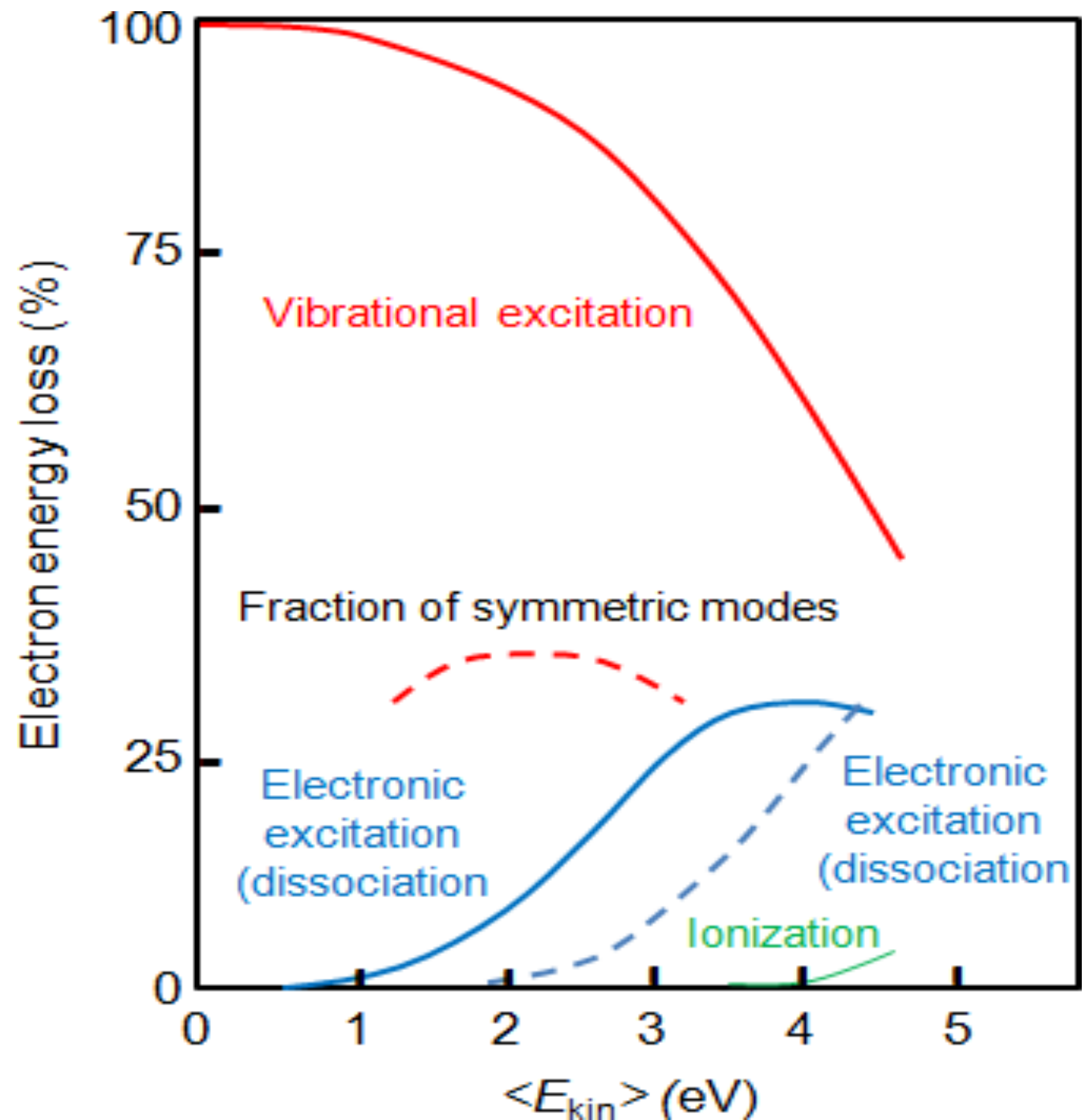
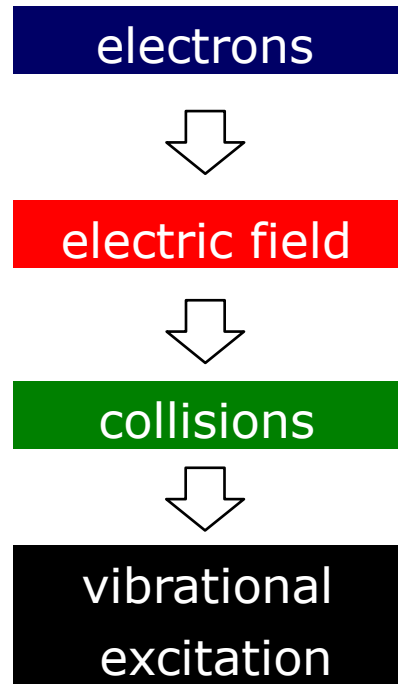
CO₂ DISSOCIATION BY STEPWISE VIBRATIONAL EXCITATION

Ladder Climbing



Annemie Bogaerts, PLASMANT,
University of Antwerp

**Electron energy loss
depends on reduced electric
field (depends on average
electron energy)**



PLASMA DISSOCIATION OF CO₂

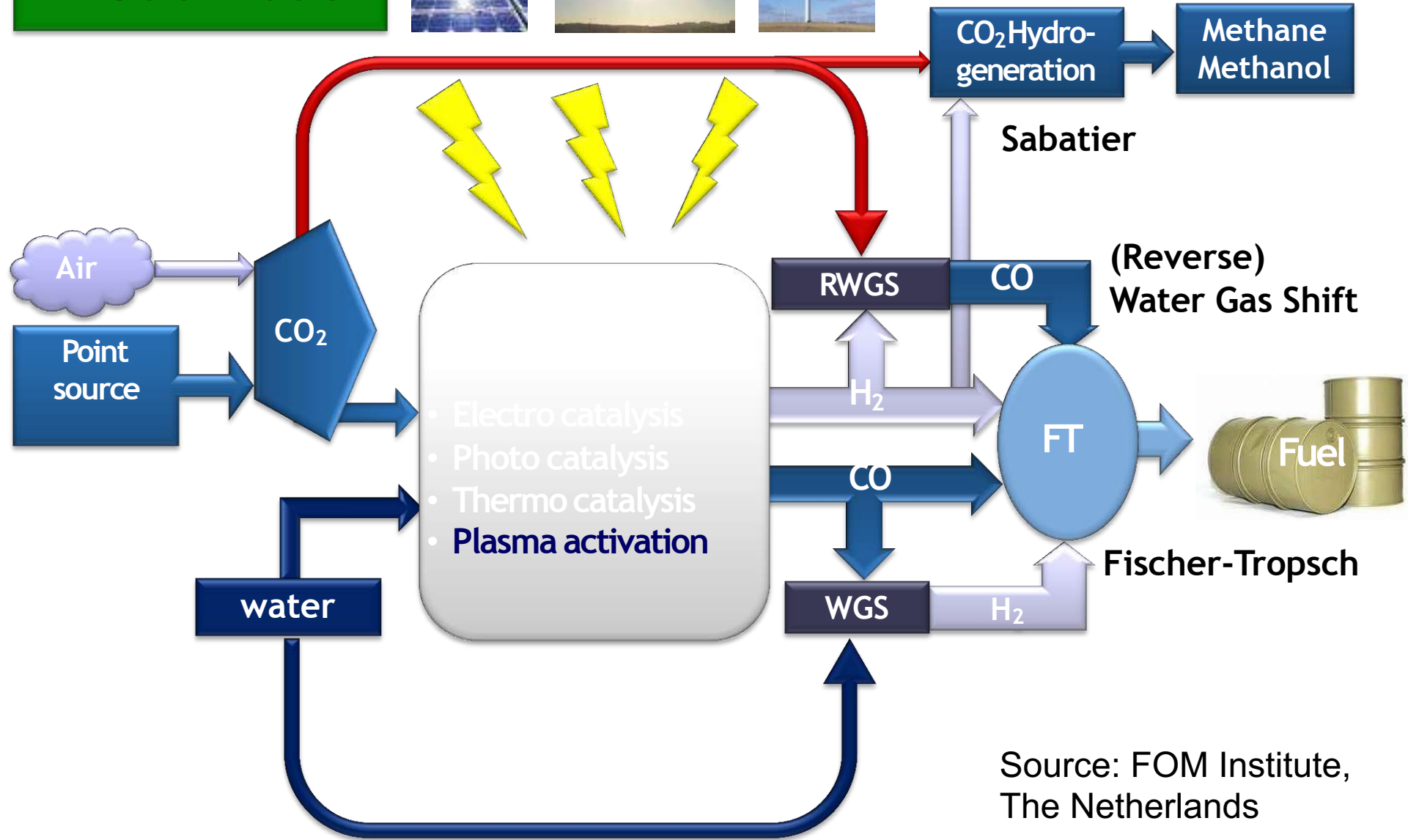
WHY PLASMA?

- **On demand capability**
- **High energy efficiency (~60% demonstrated)**
- **High power density (45W/cm³)**
- **Rapid ramping up and down (wrt high temperature SOEC)**
- **No scarce materials employed (Pt catalyst in PEM)**

	Use of rare earth metals	Renewable energy	Turnkey process	Conversion and yield	Separation step needed	Oxygenated products (e.g. alcohols, acids)	Investment cost	Operating cost	Overall flexibility
Traditional catalysis	Yes		No	High	Yes	Yes	Low	High	Low
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Solar thermo-chemical	Yes	Direct	NA	High	No	No	High	Low	Low
Photo-chemical	Yes	Direct ^a	Yes	Low	Yes	Yes	Low	Low	Low
Biochemical	No	Direct ^a	No	Medium	Yes ^d	Yes	High/low	High	Low
Plasma-chemical	No	Indirect	Yes	High	Yes ^e	Yes	Low	Low	High

Source: R. Snoeckx and A.Bogaerts, Chem. Soc. Rev., 2017,46, 5805

Grand Scheme Solar Fuels



Source: FOM Institute,
The Netherlands

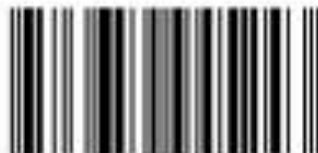
This monograph explores the pervasive role of plasma in some areas related to energy and environment. Plasma processing exploits the fact that plasmas occupy an extremely extended parameter space in density, temperature or chemical reactivity. Density can vary over 30 orders while temperature can vary over 10. Relevant time scales can vary from seconds to picoseconds. Plasmas can provide temperatures and energy densities higher than any other medium. High levels of transient and non-equilibrium conditions can prevail in plasmas. Certain plasmas can have short density and temperature scale lengths enabling fast quenching of chemical products. Plasma processing plays an important role in the realisation of modern photovoltaic devices used for converting solar energy into electricity. Atmospheric contamination from industrial and vehicular emissions is ameliorated through plasma mediated processes. The negative impact of waste in environmental degradation is reduced through plasma processes for waste to energy conversion. Hydrocarbon fuels are made carbon-lean and environmentally less damaging.



John Pucadyil



Prof. John Pucadyil served as Senior Professor and had occupied the Meghnad Saha Chair in Plasma Science and Technology at the Institute for Plasma Research in India. He has contributed significantly to the nucleation and growth of Plasma Physics and Industrial Plasma Applications in India. He lives in Kottayam, Kerala.



978-3-659-94524-3

Pucadyil

Plasma Processes for Energy and Environment

The Pervasive Role of Plasma Processing in Technologies for Clean Energy and Environment.

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